

University of Nottingham

Enterprise for Scientists

Embedding Business and Entrepreneurial Skills into the Science Curriculum

Dr. Trevor Farren Director of Business and Knowledge Exchange -School of Chemistry-



- Polymer Chemist -> Director of Development
- Interested in New Ventures & Innovation
- •Enterprise Skills Onboarding new graduates 1.5yr











University

"Embedded Industrialist" in School of Chemistry

To build and entrepreneurial culture among staff and students and increase their engagement with business and industry

Absence of Business Skills in our curriculum: not considered a worthwhile academic activity!



Provides a focus for business related activities bringing together relevant staff and projects within the School.

Ex-Industry Professionals

Post Doctoral Business Science Fellows

Students (UG/PG)

Universitu of

Secondees from Industry

Academic Staff

Fully Integrated



University of Nottingham UK | CHINA | MALAYSIA Industry Collaborations





Bringing Business and Entrepreneurial Skills into the Chemistry Curriculum

- Post Doctoral Business Science Fellowships (PD)
- From the Bench to the Bank Programme (PG)
- MSc Chemistry and Entrepreneurship (PG)
- Enterprise for Chemists (UG)
- Industry Inspired Final Year and Vacation Projects UG)
- Year in Industry and SME Placements (UG)
- Degree Apprenticeship (UG)

<u>CRODA</u>

Lubrizol





'Skills required by new chemistry graduates and their development in degree programmes, Higher Education Academy' – November 2010'





Final year optional module 60-80 students (including Physicists!)

Curriculum Design

Autumn Semester – Innovation and Entrepreneurship Spring Semester – Business Skills and Awareness

Assessment – 75% Exam, 25% Presentations

Business School Interaction





Mewburn Ellis

Universitu of Innovation and Entrepreneurship for Scientists Nottingham UK | CHINA | MALAYSIA

30/10

6th Oct

Dragons

Kick off

Chem

X-1

Mon

30th

Oct

Trevor

Farren

Phys C27

F1E4C Timetable: Semester 1

14/11

Chem X-1 / Chem X-1

9 21/11

10 28/11

Weds/

Thurs 30

Nov 01st

Dec

Dragon's

Den

11

12

11/12

Feedback

04/12

Dragons

mentoring

Den

Dragons

mentoring

Den

4 Taught Sessions on Innovation

2 03/10

Mon

3rd

Oct

Steve

Howdle

Pm Phys

C27

3

4

LECTURE 23rd Oct

Boots

Phys C27

5 123/10 -

17/10

No

10/10

Mon

10th

Oct

Simon

Mosey

Phys

 c_{22}

- A Practical Overview
- **Business School Model** •
- **Company Model**

University Model



Science Concentrates

m1

INFECTIOUS DISEASE Zika virus needs a neural stem cell **protein to replicate**

The Zika virus outbreak in the Americas has revealed the devastating effects of the pathogen on developing fetal brains. Researchers are still trying to tease apart the precise mechanisms the virus uses to cross the placental barrier and cause microcephaly in babies. A research team led by Fanni Gergely at the University of Cambridge is now reporting that the virus relies on a protein called Musashi 1 found in fetal neural stem cells to replicate (Science 2017, DOI: 10.1126/ science.aam9243). The higher the level of Musashi 1 in a neural stem cell, the more Zika virus is produced and the faster the cell dies. When the researchers reduced levels of Musashi 1 in stem cells, the pathogen could not replicate quickly. The team notes that the virus may target neural stem cells because the protein does not get produced in mature neurons. To confirm that diversion of Musashi 1 away from its typical role in brain development might lead to microcephaly, the team tracked down a family with two children who were both born with microcephaly independent of the Zika outbreak. The children had mutations in the gene for Musashi 1. Blocking the interaction between Zika RNA and Musashi 1 might reduce the pathogen's effects on a developing brain, the researchers suggest.-SARAH EVERTS

FLECTRONIC MATERIALS

Liquid metal 'does a solid' for semiconductors

Growing high-quality semiconducting crystals isn't easy. It typically takes high temperatures, highly reactive precursors, and extensive equipment. Electrochemical deposition could simplify the process, but attempts using conventional solvents vield "amorphous junk," says Stephen Maldonado of the University of Michigan. Working with colleagues at Ohio State University, Maldonado's team has found that a liquid metal can POLYMERS Artificial vesicle regulates glucose levels in mice

Chinese researchers hope that their new polymersome-a polymer vesiclehas the potential to regulate glucose levels in diabetes without insulin. Yufen Xiao, Hui Sun, and Jianzhong Du of Tongji University make their polymersomes of a sugar-containing copolymer with a poly(ethylene oxide) tail (J. Am. Chem. Soc. 2017, DOI: 10.1021/jacs.7b03219). Sugar side chains in the co-

polymer interact with the sugar-binding protein conalucose canavalin A, which is then level immobilized inside the vesicle by cross-linking. Low The polymersome soaks glucose up glucose when the surrounding concentration is high and releases it when the concentration is low because concanavalin A has a higher affinity for free glucose than for the sugar side chain A glycopolymersome self-assembles with attached to the vesicle.

a copolymer as the vesicle membrane and The researchers tested a poly(ethylene oxide) tail on the surface. this glucose sponge in Concanavalin A (pink) associates with the mice with chemically copolymer's sugar side chains (brown). At high induced elevated blood glucose concentrations, the glycopolymersome absorbs free glucose (black), which displaces blood glucose level of these mice dropped and the sugar side chains. As the glucose lavel stayed at normal levels drops, the reverse happens. for at least 36 hours after

injection of the vesicles. "As a concept this is rather clever, and the demonstration in mice looks promising," says Cameron Alexander, an expert on polymer therapeutics at the University of Nottingham. "The caveat is that these are complex systems, and a lot of development and tolerability studies will be needed before these are anywhere close to being used in humans."--CELIA ARNAUD

eGaIn. This liquid metal forms an internet help electrochemistry conquer this shortcoming (J. Am. Chem. Soc. 2017, DOI: 10.1021/ diate layer between the solution and a solid jacs.7b01968). The researchers started with an aqueous germanium oxide solution and introduced eutectic gallium-indium, or



glucose levels. The

A new electrochemical process puts semiconductors at researchers' fingertips (left). The process creates germanium thin films (green) on silicon (orange) with specks of gallium (white), as shown by electron microscopy and X-ray analysis (right).

silicon substrate. Germanium reduced in water can cross the interface into the eGaIn and migrate to the silicon. By tuning the thickness of the eGaIn laver.

the researchers ensure that germanium accumulates into a high-quality, crystalline film. This low-cost, benchtop approach works at room tempera ture, unlike conventional semiconductor growth, Maldonado says. The group is exploring liquid metal solvents further, hop

ing to inspire others to rethink how they make semiconductor circuits and devices .- MATT DAVENPORT

UNE 5 2017 | CEN.ACS.ORG | CAEN Q

Team Exercise (4)

3 weeks

3 'flipped classroom' sessions

University of Nottingham UK | CHINA | MALAYSIA The Business Pitch

